THE SYNTHESIS OF DL-TETRAHYDROANHYDROAUCUBIGENONE

Heitaro OBARA, Hiroshi KIMURA, Jun-ichi ONODERA, and Masanobu SUZUKI

Department of Applied Chemistry, Faculty of Engineering,

Yamagata University, Yonezawa-shi, 992

The racemate of tetrahydroaucubigenone (III), one of the derivatives of aucubin (I), was synthesized by the photo-irradiation of 2-(2'-nitrosoxyethyl)-7-oxabicyclo[3,3,0] octan-3-one (XV) which was obtained from 2-ethoxycarbonyl-7-oxabicyclo[3,3,0] octan-3-one (VIII) through seven steps.

Tetrahydroanhydroaucubigenone (III), mp 77-78°C, $(\alpha)_D^{17}$ - 41.3° (ethanol), an oxidation product of tetrahydroanhydroaucubigenin (II) which was derived from aucubin (I), has been first obtained by Fujise²⁾ in 1953. The synthesis of III, which is thought to be an important compound for the determination of I, has not been accomplished yet, although some derivatives of I have been obtained.³⁾

Our previous observation that 2,8-dioxabicyclo [3,3,0] octane (V) was prepared by photochemical reaction of 2-(tetrahydrofuryl)ethylnitrite (IV) along with an acetaldehyde derivative (VI) and an ethanol one (VII)⁴⁾ promoted us to investigate this phenomenone in the 7-oxabicyclo [3,3,0] octan-3-one series.

In this communication we wish to describe the photochemical preparation of DL-III by the following scheme.

2-Ethoxycarbonyl-2-ethoxycarbonylmethyl-7-oxabicyclo [3,3,0] octan-3-one (IX), was obtained in a 62% yield with a small amount of 0-alkylated compound by the alkylation of the sodio derivative of cis-2-ethoxycarbonyl-7-oxabicyclo [3,3,0] - octan-3-one (VIII)⁵⁾ with ethyl bromoacetate in toluene. Hydrolysis and decarboxylation of IX with 6 N hydrochloric acid, followed by esterification, afforded 2-ethoxycarbonylmethyl-7-oxabicyclo [3,3,0] octan-3-one (X) in a 55% yield, bp 129-131°C/2mmHg, 2,4-dinitrophenylhydrazone, mp 144°C. After protection of the carbonyl group of X with ethylene glycol in the presence of p-toluenesulfonic acid in benzene, the ethylene-ketal (XI) was reduced into the alcohol (XII) in a 89% yield with lithium aluminium hydride in tetrahydrofuran. Deketalization of XII by treatment with dilute hydrochloric acid, or p-toluenesulfonic acid in acetone,

afforded a mixture of 2-(2'-hydroxyethyl)-7-oxabicyclo [3,3,0] octan-3-one (XIII) and its intramolecular hemiketal (XIV) in a 76% yield, 2,4-dinitrophenylhydrazone, mp 154°C. The hemiketal (XIV) is easily converted into keto-alcohol (XIII) by treatment with dilute hydrochloric acid. 2-(2'-Nitrosoxyethyl)-7-oxabicyclo-[3,3,0] octan-3-one (XV), bp 100-102°C/2mmHg, a pale green viscous oil, was obtained in a 30% yield from the keto-alcohol (XIII) by treatment with a mixture of sodium nitrite and dilute sulfuric acid.

The irradiation of XV (3g) was carried out in a benzene solution under a nitrogen atmosphere at room temperature for 15 hr using a high-pressure 500 W mercury arc. After irradiation, the benzene was removed in vacuo, and the residue was chromatographed on a column of silica gel with carbontetrachloride-ethyl acetate (3:1) to give DL-III (14mg), colorless prisms, mp 57-57.5°C, (recrystallized from petroleum ether-benzene). The ir (Fig. 1), nmr, and mass spectra of this compound were completely identical with those of the natural sample (III).

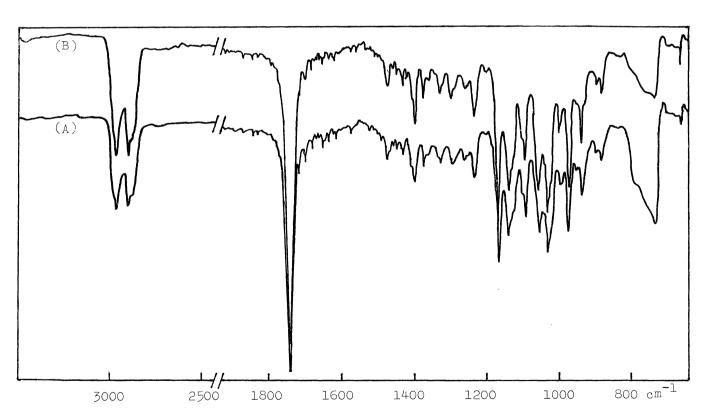


Fig. 1. Infrared spectra of DL-tetrahydroaucubigenone (A) and III (B). (in CCl_4).

References

- S. Fujise, H. Obara, and H. Uda, Chem. & Ind., 1960, 289; S. Fujise, H. Uda, and H. Obara, Nippon Kagaku Zasshi, 81, 677 (1960); S. Fujise and H. Kojima, ibid., 82, 1110 (1961); H. Uda, ibid., 81, 1865 (1960); J. Grimshaw and H. R. Juneja, J. Chem. Soc., 1961, 5194; M. W. Wendt, W. Haegele, E. Simonitsch, and H. Schmid, Helv. Chim. Acta, 43, 1440 (1960); W. Haegele, K. Kaplan, and H. Schmid, Tetrahedron Letters, 1961, 110.
- 2) S. Fujise, Nippon Kagaku Zasshi, <u>74</u>, 725 (1953).
- H. Obara, Nippon Kagaku Zasshi, <u>81</u>, 1871 (1960); K. Kurosawa and S. Fujise,
 ibid., <u>83</u>, 329 (1962); K. Kurosawa and S. Fujise, Chem. & Ind., <u>1963</u>, 1688;
 K. Kurosawa, H. Obara, and H. Uda, Bull. Chem. Soc. Japan, <u>39</u>, 530 (1966).
- 4) H. Obara, H. Kimura, and J. Onodera, The Abstract of the 22nd Annual Meeting of the Chemical Society of Japan (1969), III, p. 1354.
- 5) H. Obara, S. Kumazawa, J. Onodera, and H. Kimura, Nippon Kagaku Kaishi, <u>1974</u>, 2380.

(Received January 6, 1975)